# **ORIGINAL RESEARCH ARTICLE**

# Bio stimulation of indigenous microorganisms with Gomeya: a bioremediation technique

# \*1,2 I. A. Anuoluwa and <sup>2</sup>A. A. Ogunjobi

# Affiliation

<sup>1</sup>Department of Biological Sciences, Faculty of Sciences, University of Medical Sciences, Ondo State, Nigeria

<sup>2</sup>Environmental Microbiology and Biotechnology Unit, Department of Microbiology, Faculty of Science, University of Ibadan

# \*For Correspondence

E-mail:dunnibright@yahoo.com, ianuoluwa@unimed.edu.ng, Tel: +234-8034336722

# ABSTRACT

The burden of heavy metals in the environment can be reduced using organic amendment stimulated bacterial remediation. This study employed cattle manure slurry stimulated bacterial inoculum to treat heavy metals-contaminated soil. Samples of contaminated soil and cattle manure were collected from the area surrounding a steel rolling mill and a commercial animal pen respectively. Bacteria were isolated using pour plate technique; identified using various biochemical tests and screened for resistance to heavy metal salts by incorporating heavy metal salts into agar plates. The contaminated soil and manure slurry were analysed for heavy metals and then sterilised separately. Five kilogram of the sterilised contaminated soil was weighed and mixed with 100g of sterilized cow dung slurry and aseptically packed into plastic nursery bags. Bacterial samples showing high tolerance to heavy metal salts were introduced into the bags singly and as a consortium for bioremediation exercise. Thirty-six bacterial isolates were obtained from the contaminated soil. Chemical analysis revealed that the soil was heavily contaminated especially with lead and chromium with concentrations of 1505.1-2333.6 and 1526.0-1678.7 mg/kg, respectively. Alcaligenes faecalis, Pseudomonas azotoformans and Bacillus mycoides exhibited high tolerance to heavy metals salt and were selected for bioremediation. Post bioremediation analysis of the soil samples revealed a reduction in the concentration of heavy metals concentration with major reduction in the concentration of chromium in groups treated with P. azotoformans. Biostimulation of microorganisms with organic amendment effectively remediated heavy metals contaminated soil and can be employed in the treatment of such contaminated environments.

**Biological:** Microbiology. Keywords: Cattle manure slurry, biomagnification, stimulated bacterial inoculum

# **INTRODUCTION**

Some important heavy metals such as cadmium, copper, lead, chromium and mercury are also important environmental contaminants and they are found in high concentration especially in areas with high anthropogenic activities (Suruchi and Khanna, 2011). Though some of these heavy metals are also soil micronutrients; the extent of soil pollution by these heavy metals and base metal ions is alarming. As a result of pollution, it has been observed that the larger the urban area, the lower the quality of the environment (Eddy, 2004a). The various sources through which heavy metals are released into the soil environment include but are not limited to natural means such as emissions from volcanoes, transport of continental dust and the weathering of metal-enriched rocks (Ernst, 1998) or as a result of various anthropogenic activities such as exploration of mines and smelters, the application of manures, fertilizers, metal based pesticides and metal-enriched sewage sludge in agriculture, combustion of fossil fuels, metallurgical industries, military training, manufacturing, usage and disposal of electronics.

**Heavy metal** uptake by plants grown on polluted soils has been studied to a considerable extent (Suruchi and Khanna, 2011; Navarro et al., 2008; Dixit et al., 2015; Sukreeyapongse et al., 2002, Yusuf et al., 2003). Heavy metal uptake via roots from contaminated soils and surface water, and direct deposition of heavy metal contaminants from the atmosphere unto plant surfaces can lead to contamination of plant by heavy metals. When these metals exceed the physiological demand of plant, they may not only be toxic to the plants, but oftentimes enter into the food chain, become biomagnified and pose serious health concern to humans (Sugiyama, 1994; Odoh and Kolawole, 2011). The biotoxic effects of heavy metals on plants depend upon the concentrations and oxidation states of heavy metals, its source and mode of deposition (Duruibe et al., 2007).

Though various conventional technologies such as chemical oxidation, precipitation, ion exchange, soil washing, incineration, solidification and stabilization are usually employed in the remediation of heavy metals contaminated soil (FRTR, 2000; Gomes et al., 2012) biological treatment of heavy metal-contaminated soil is often more attractive than direct chemical or physical treatment. One of the most promising technological approaches to the problem of heavy metal contamination in the environment is bioremediation. This can be attributed to the ability of the microorganisms to directly sequester contaminants rather than merely transferring them from one medium to another (USEPA, 1995). Some organisms which have been involved in bioremediation processes include Pseudomonas aeruginosa, P. ambigua, P. fluorescens, other species such as Bacillus cereus, B. subtilis, E. coli (ATCC 33456), Achromobacter Eurydice, Micrococcus roseus, Enterobacter cloacae, Desulfovibriode sulfuricans and D. vulgaris. Shewanella algae BrY-MT have been reported to be effective in bioremediation of various contaminants (Guha et al., 2012; Camargo et al., 2003). Comparing the effectiveness of conventional methods of metal removal with biological methods of metal removal, it has been observed that the use of biomass of microorganisms has distinct advantages over conventional methods due to the fact that they are highly selective and cost effective, have diversity of active binding site (Ahluwalia and Goyal, 2007; Green-Ruiz et al., 2008). For instance, microorganisms can assimilate heavy metals actively (bioaccumulation) and/or passively (adsorption) (Hussein et al., 2001). The bacterial cell walls, which consist mainly of polysaccharides, lipids and proteins, offer many functional groups that can bind heavy metal ions, and these include carboxylate, hydroxyl, amino and phosphate groups (Randhawa and Kullar, 2011). Algae, fungi, yeast, protozoa and bacteria have been employed in the removal of heavy metals from industrial waste waters by using the microorganisms in whole and/or using products of their metabolism such as enzymes and biosurfactant (Congeevaram et al., 2007; Özdemir and Kılınc, 2012; Özdemir et al., 2012).

Cattle manure slurry/gomeya usually referred to as a waste product can enhance the degradation of contaminants in the environment. Cattle dung slurry is a cheap and easily available rich source of organic amendment. It is a mixture of cattle dung and urine in a ratio of around 3:1 respectively (Randhawa and Kullar, 2011). According to Adedokun and Ataga (2007), soil amendments or The Proceedings of the Nigerian Academy of Science

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additives are needed to increase the activities of microbes and for effective bioremediation of polluted soil.

*Corchorus olitorius* commonly called jute or Jew mallow belongs to the Tiliaceae family. The choice of *C. olitorius* for this study is based on the fact that it is one of the most popular vegetables in every home; hence it is grown in nearly all home gardens, market gardens near the city and truck gardens around the world (Aluko et al., 2014).

The aim of this study was to bioremediate heavy metals contaminated soil using organic amendment stimulated bacterial remediation. This study was therefore designed to study the ability of bacteria isolated from heavy metal-contaminated soil to effectively remove heavy metals in order that the bioremediated soil can support plant growth.

# **Materials and Methods**

# Sample collection

The study location was densely contaminated with heavy metals contained in effluent released as a result of the activities of a steel rolling company in South-western Nigeria. Consequently, plants grown in the vicinity of the company could not thrive which had an impact on the livelihood of people living in that area as majority of them were peasant farmers. The study location was visited during the dry and rainy season to carry out *in situ* analysis in order to obtain background knowledge of the seasonal variations in the physical and chemical properties of the location. Physical observation was done to observe changes in physical characteristics such as colour, texture, odour and deposition of effluents from the rolled steel industry which were considered as indicators for pollution. Soil sample was collected from different points of the location using a soil auger. The soil was transported to the Department of Microbiology, University of Ibadan where microbiological and chemical analysis were conducted within 24 h of collection.

# Analysis of heavy metal-contaminated soil

The soil samples collected were thoroughly mixed using a hand trowel to obtain a composite sample before subjecting it to various analyses. Soil hydrogen ion concentration (pH 1:1 H<sub>2</sub>O) was determined using a glass electrode pH meter (Hanna instruments HI2210) following the methods described by Bates (1954). The exchangeable acidity was determined using the KCl extraction method following the method of Mclean (1965), the organic matter in the soil sample in the form of carbon was determined using the Walkey-Black wet oxidation method as described by Page (1982), the total nitrogen in the composite soil sample was determined using the macro-Kjeldahl method as described by Page (1982), the phosphorus in the composite soil sample was analysed using the vanado-molybdate method (AOAC, 2012). The calcium and magnesium content of the soil was determined using methods described by Mehlich (1953) and Watanabe and Olsen (1965). The concentration of heavy metals such as cadmium, iron, copper, lead, chromium, zinc, nickel, manganese and cobalt present in the soil sample was determined using the wet digestion procedure (SSSA, 1971). This was carried out by weighing 0.5 g of the 0.5 mm sieved soil into a 100 mL Berzellius beaker, 5 mL HNO3 and 2 mL HClO4 was added and covered with a watch glass. This was digested in a fume cupboard by heating it to a final volume of 3 to 5 mL. Ten to fifteen millilitres of water was added and the digest solution was filtered through an acid washed filter paper into a 50 mL volumetric flask. It was diluted to volume with deionized water and the filter paper was washed with water. The filtrate was used to determine the concentration of heavy metals present in the sample using Buck Scientific 210/211 VGP Atomic Absorption Spectrophotometer (AAS).

#### **Isolation of microorganisms**

The determination of the total viable bacteria count (TVBC) was carried out in triplicates. The agar medium and the diluents used were sterilized at 121°C for 15 minutes. One gram of the thoroughly mixed composite soil samples was suspended in 9 mL of sterile distilled water and serially diluted (Olutiola et al., 2000). One millilitre of appropriate dilutions was inoculated into sterile Petri dishes and already prepared and cooled nutrient agar (Lab M, United Kingdom) was added to it using the pour plate technique as described by Olutiola et al. (2000). Inoculated plates were incubated at 37°C for 24 h after which distinct bacteria colonies were counted. Morphologically distinct bacteria colonies were subcultured by streaking on fresh nutrient agar plates until pure bacteria colonies were obtained. Pure cultures of each bacteria strain were stored on nutrient agar slants at 4 °C for further studies. Pure bacterial isolates were subjected to various biochemical tests to aid their identification.

# Molecular Characterisation of Bacterial Isolates 16S rRNA based identification

Isolation of 16S rRNA gene of the bacterial isolates were carried out using QIAamp DNA Mini Kit (250) cat no 51306 after which the sequences were amplified using Applied Biosystems Thermocycler, model 9800. Sequencing of the 16S rRNA was carried out using a 16-well Applied Biosystems sequencing plate following the manufacturer's instructions. The obtained sequences of bacterial 16S rRNA were analysed using Sequence Scanner (Applied Biosystems) software and the 16S rRNA sequence contigs were generated using Chromas Pro. The online program BLASTn was used to find out the related sequences with known taxonomic information in the databank at NCBI website (http://www.ncbi.nlm.nih.gov/BLAST) to accurately identify the bacterial strain. The data obtained from the molecular characterization was used in constructing a phylogenic tree for the bacterial isolates and also submitted to DNA Data Bank of Japan (DDBJ) for accession numbers.

# **Phylogenetic Analyses of bacterial strains**

The 16S rRNA gene sequences obtained from the GenBank database of the National Centre for Biotechnology Information (NCBI) were aligned using the Molecular Evolutionary Genetics Analysis (MEGA) software version 6 following the method described by Hall (2013) and Tamura et al. (2013). The evolutionary history of the bacterial isolates was inferred using the Neighbor-Joining method as described by Saitou and Nei (1987). The tree was drawn to scale, with branch lengths in the same units as those of the evolutionary distances used to infer the phylogenetic tree. The evolutionary distances were computed using the Maximum Composite Likelihood method (Tamura et al., 2004) and in the units of the number of base substitutions per site.

# Determination of the susceptibility of bacterial isolates to heavy metal salts

The susceptibility of the bacterial isolates to increasing concentration of chromium, cadmium, lead, copper, cobalt, nickel and zinc was determined quantitatively using the agar diffusion method following the method described by Narasimhulu *et al.* (2010). Concentration of heavy metals in nutrient agar medium was gradually increased from 100-500  $\mu$ g/mL. The screening was done by streaking a 24 h old culture of the test organism on nutrient agar plate supplemented with 100  $\mu$ g/mL of the salt of the heavy metals of interest and was incubated for five days. Isolates that grew at this concentration were subcultured to nutrient agar plates supplemented with higher concentration of the heavy metal salts until 500  $\mu$ g/mL concentration of heavy metal at increasing level of 50 $\mu$ g/mL was attained. Isolates which were observed to have high tolerance to heavy metal The Proceedings of the Nigerian Academy of Science

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salts were maintained on agar slants and stored at 4 °C to be used for bioremediation of the contaminated soil samples.

# Sterilization of soil sample

Composite soil samples collected from the heavy metal-contaminated steel rolling site were air dried in the sunlight for a day and then sieved using a 0.5 mm nylon mesh sieve. The soil sample was then sequentially sterilized using hot air oven at 105 °C for one hour, after which it was aseptically packaged by weighing 5 kg into sterile polythene bags for the planting exercise as described by Saeed and Rafique (1980) and Iqbal *et al.* (2011). To check for sterility of the soil sample, the soil sample was subjected to the method used for isolation of microorganisms as described above.

# **Collection of Corchorus olitorius seeds**

For the purpose of this study, seeds of *Corchorus olitorius* already treated with scarification method (in order to break the dormancy) were obtained from Agronomy Department, Faculty of Agriculture, University of Ibadan, Nigeria.

# Bioremediation of contaminated soil sample

Five kilogram of already sterilized soil sample were aseptically weighed and mixed with one hundred gram of sterilized cow manure slurry. This was then packed into perforated polythene bags (nursery bags) to be used for the bioremediation and planting exercise. The bioremediation and planting exercises were conducted in a screen house. Working solution for the bioremediation exercise was prepared using a modified method of Ayotamuno *et al.* (2009) by inoculating each of the bacterial isolates into peptone water broth and incubating until a cell density of  $7.6 \times 10^{11}$  cfu/mL was obtained, however, for bioremediation exercise using mixed culture, the working solution was allowed to reach a cell density of  $1.5 \times 10^{12}$  cfu/ml as described by Okparanma *et al.* (2009). Twenty millilitres of the working solution of the bacterial isolate was pipetted into each experimental setup. The experiment was laid out in a Completely Randomized Design (CRD) having eight treatments with five replicates making a total of forty pots.

# **Treatments:**

- B Sterilized soil treated with *Pseudomonas azotoformans* (LC349894.1) and manure
- C Sterilized soil treated with *Bacillus mycoides* (LC349697.1) and manure
- AB Sterilized soil treated with consortia of *Alcaligenes faecalis* (LC349889.1) and *Pseudomonas azotoformans* (LC349894.1) and manure
- AC Sterilized soil treated with consortia of *Alcaligenes faecalis* (LC349889.1) and *Bacillus mycoides* (LC349897.1) and manure
- BC Sterilized soil treated with consortia of *Pseudomonas azotoformans* (LC349894.1) and *Bacillus mycoides* (LC349897.1) and manure
- ABC Sterilized soil treated with consortia of *Alcaligenes faecalis* (LC349889.1), *Pseudomonas azotoformans* (LC349894.1) and *Bacillus mycoides* (LC349897.1) and manure control Sterilized soil alone (without bacterial inoculums and manure).

A Sterilized soil treated with *Alcaligenes faecalis* (LC349889.1) and manure

# Planting of Corchorus olitorius

Planting of *C. Olitorius* was done immediately after introducing the bacterial isolates. The pots were watered twice daily (100 mL/time) for the first two weeks of planting and later reduced to once daily so as to prevent the leaching of nutrient from the treatments. This was done till the experiment was terminated seven weeks after planting (WAP). Plant height was measured from the rhizoplane to the apical tip of the plant in centimetres starting from 2WAP till the seventh week when the experiment was terminated, stem diameter measured using a vernier caliper below the first nodes of the plant and number of leaves obtained by counting the number of leaves per plant manually Elings (2000).

#### Analysis of harvested C. olitorius

At the end of the experiment, the plants were harvested by uprooting the plantings and pooled together per treatment group. The roots were washed to remove attached soil debris and then subjected to various analyses. The wet and dry weight, moisture content, dry matter, % crude protein, % ash, % ether extract (fat), % crude fibre were determined using standard analytical methods as described by AOAC (2012). The heavy metal content of the plant samples was determined following the method described by SSSA (1971) using Buck Scientific 210/211 VGP Atomic Absorption Spectrophotometer (AAS).

#### Analysis of the bioremediated soil samples

Upon termination of the experiment and harvest of the plants, the five soil replicates in each treatment group were pooled together and mixed well in order to obtain a composite sample which was analysed for pH, nitrogen, organic carbon, exchangeable acidity, available phosphorus, calcium, potassium, sodium, magnesium, manganese, iron, copper, zinc, lead, cadmium, chromium, cobalt and nickel content following the methods described earlier.

#### **Data Analysis**

Data obtained were analysed and reported as mean  $\pm$  standard deviation of five measurements and analysed using univariate analysis of variance and Duncan Post Hoc test to determine significant differences (p $\geq$ 0.05) between treatments using Statistical Package for Social Science Research version 17 (SPSS).

#### Results

The collected soil samples were dark brown in colour with a characteristic choking odour which is peculiar to heavy metal contaminated sites. Table 1 shows the result of the initial *in situ* analysis carried out to determine the physical, chemical and heavy metals properties of the composite soil samples collected during the rainy and dry season. It was observed that the soil from the study site had higher physical, chemical and heavy metal content during the dry season than the rainy season. For instance, heavy metals such as cadmium, lead, cobalt, nickel and chromium had concentrations of 3.0, 2333.6, 13.7, 40.6 and 1678.7 mg/kg, respectively during the dry season and concentrations of 0.5, 1505.5, 10.5, 31.5 and 1526.0 mg/kg, respectively during the rainy season.

Parameters	Dry season	Rainy season			
Ph	6.5	7.0			
Exchangeable Acidity(meq/100g)	0.4	0.4			
Mineral content					
Total organic carbon (TOC) (g/kg)	42.4	40.8			
Total Nitrogen (g/kg)	4.0	3.9			
Available Phosphorus(mg/kg)	14.0	13.3			
Ca (cmol/kg)	2.6	2.7			
K (cmol/kg)	0.3	0.4			
Na (cmol/kg)	0.5	0.3			
Mg (cmol/kg)	0.5	0.4			
Heavy metals (mg/kg)					
Mn	99.4	98.5			
Fe	24.8	23.8			
Cu	2.8	1.0			
Zn	2.6	1.4			
Cd	3.0	0.5			
Pb	2333.6	1505.5			
Со	13.7	10.5			
Ni	40.6	31.5			
Cr	1678.7	1526.0			

# Table 1: Physical and chemical properties of the heavy metals contaminated soil sampleduring dry and rainy seasons

Thirty-six bacterial isolates were obtained from the contaminated soil sample and they were distributed as follows: *Pseudomonas* sp (52.77%), Proteus *mirabilis* (13.89%), *Alcaligenes faecalis* (13.89%), *Enterobacter* sp (8.33%), *Providencia* sp (5.56%) and *Bacillus* sp (5.56%).

However only three of these isolatesviz: *Alcaligenes faecalis* (LC349889.1), *Pseudomonas azotoformans* (LC349894.1) and *Bacillus mycoides* (LC349897.1) exhibited high tolerance to heavy metal salts, therefore the three bacterial isolates were selected for bioremediation exercise. Figures 1-3 shows the phylogenetic tree constructed for each of the isolates based on the molecular data obtained when they were identified using molecular characterization (isolates appear in bold format).

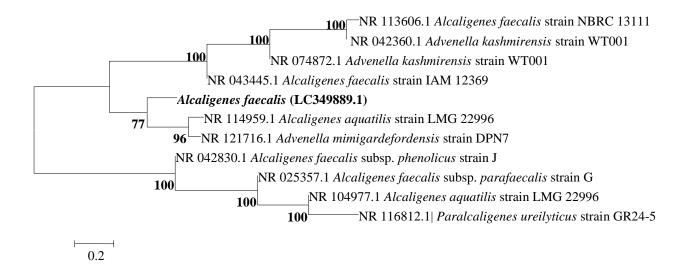


Figure 1: Evolutionary relationship of Alcaligenes faecalis (LC349889.1)

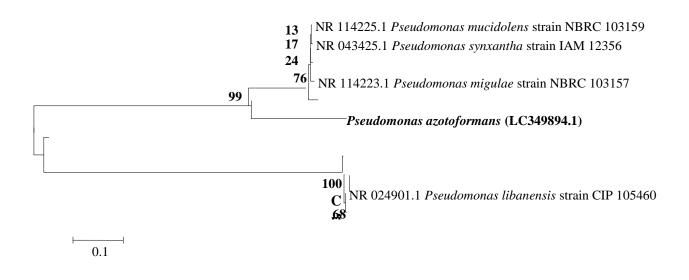


Figure 2: Evolutionary relationship of *Pseudomonas azotoformans*(LC349894.1)

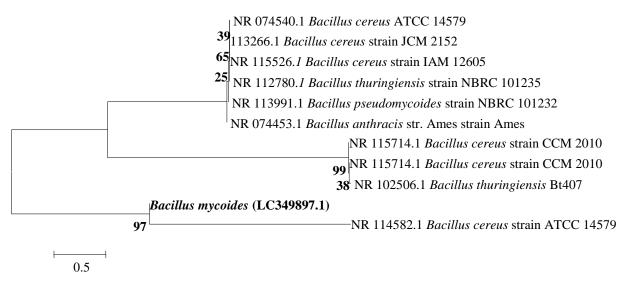
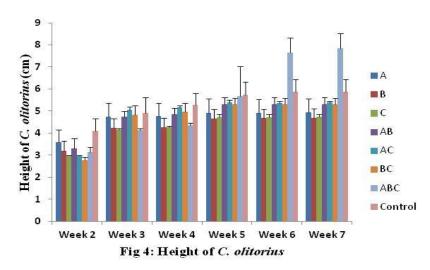


Figure 3: Evolutionary relationship of *Bacillus mycoides* (LC349897.1)

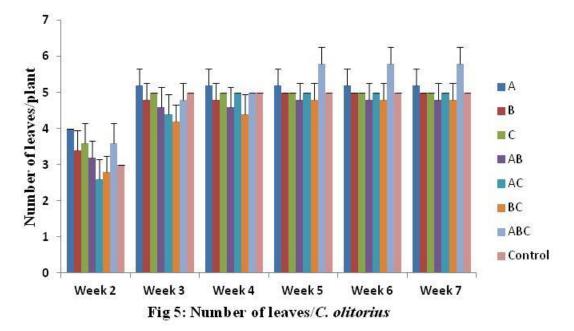
The result obtained for the monitored agronomic parameters is as shown in Figures 4-6. There was significant difference (at  $p \ge 0.05$ ) in the plant height of the treatments starting from 2WAP to 7WAP when the experiment was terminated as shown by Figure 4. It was observed that the plant response in the various treatment groups varied in their performance. It was observed that *C. Olitorius* harvested from soil bioremediated with a mixed culture of *Pseudomonas azotoformans* (LC349894.1)and *Bacillus mycoides* (LC349897.1) had the least height of 2.76±0.15 cm at 2 WAP, however at 6 WAP, the highest height was observed in *C. Olitorius* harvested from soil bioremediated from soil bioremediated with a consortia of *Alcaligenes faecalis* (LC349889.1), *Pseudomonas azotoformans* (LC349894.1) and *Bacillus mycoides* (LC349897.1) with a height of 7.84±0.69 cm, while the least height was observed in *C. olitorius* harvested from soil bioremediated from soil bioremediated with *Pseudomonas azotoformans* (LC349894.1) having a height of 4.68±0.41 cm, and this trend was maintained till the experiment was terminated. The control had the highest height of 4.10±0.55 cm as at 2WAP and at 7WAP it had a height of 5.86±0.57 cm.



**Keys:** A- *C. Olitorius* harvested from sterilized soil bioremediated with *Alcaligenes faecalis* (LC349889.1), B- *C. Olitorius* harvested from sterilized soil bioremediated with *Pseudomonas* 

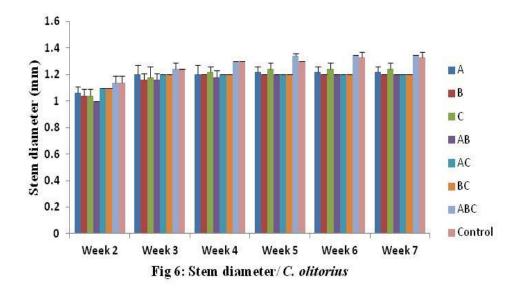
The Proceedings of the Nigerian Academy of Science Volume 13, No 1, 2020 azotoformans(LC349894.1), C- C. Olitorius harvested from sterilized soil bioremediated with Bacillus mycoides (LC349897.1), AB- C. Olitorius harvested from sterilized soil bioremediated with mixed culture of Alcaligenes faecalis (LC349889.1) and P. Azotoformans (LC349894.1), AC-C. Olitorius harvested from sterilized soil bioremediated with mixed culture of Alcaligenes faecalis (LC349897.1), BC- C. Olitorius harvested from sterilized soil bioremediated from sterilized soil bioremediated with mixed culture of Alcaligenes faecalis (LC349897.1), BC- C. Olitorius harvested from sterilized soil bioremediated with mixed culture of P. Azotoformans (LC349894.1) and Bacillus mycoides (LC349897.1), ABC- C. Olitorius harvested from sterilized soil bioremediated with consortia of Alcaligenes faecalis (LC349889.1), Pseudomonas azotoformans (LC349894.1) and Bacillus mycoides (LC349897.1), Control- C. Olitorius harvested from sterilized soil bioremediated with consortia of Alcaligenes faecalis (LC349897.1), Control- C. Olitorius harvested from sterilized soil alone.

The results obtained for number of leaves/plant revealed that there was significant difference (at  $p\leq0.05$ ) in the number of leaves per plant among the treatments starting from 2WAP as shown by Figure 5. *C. Olitorius* harvested from soil bioremediated with *Alcaligenes faecalis* (LC349889.1) had the highest number of leaves/plant from 2WAP till 4WAP having an average of  $4.00\pm0.00$  and  $5.20\pm0.45$ , respectively but by 5WAP till the 7WAP when the experiment was terminated, *C. Olitorius* harvested from soil bioremediated with a consortia of *Alcaligenes faecalis* (LC349889.1), *Pseudomonas azotoformans* (LC349894.1) and *Bacillus mycoides* (LC349897.1) had the highest number of leaves/plant with an average of  $5.80\pm0.45$ .



**Keys:** A- *C. olitorius* harvested from sterilized soil bioremediated with *Alcaligenes faecalis* (LC349889.1), B- *C. olitorius* harvested from sterilized soil bioremediated with *Pseudomonas azotoformans* (LC349894.1), C- *C. olitorius* harvested from sterilized soil bioremediated with *Bacillus mycoides* (LC349897.1), AB- *C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *Alcaligenes faecalis* (LC349889.1) and *P. azotoformans* (LC349894.1), AC- *C. olitorius* harvested from sterilized soil bioremediated from sterilized soil bioremediated with mixed culture of *Alcaligenes faecalis* (LC349889.1) and *P. azotoformans* (LC349894.1), AC- *C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *Alcaligenes faecalis* (LC349897.1), BC- *C. olitorius* harvested from sterilized soil bioremediated from sterilized soil bioremediated with mixed culture of *P. azotoformans* (LC349894.1) and *Bacillus mycoides* (LC349897.1), ABC- *C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *P. azotoformans* (LC349894.1) and *Bacillus mycoides* (LC349897.1), ABC- *C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *P. azotoformans* (LC349894.1) and *Bacillus mycoides* (LC349897.1), ABC- *C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *P. azotoformans* (LC349894.1) and *Bacillus mycoides* (LC349897.1), ABC- *C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *P. azotoformans* (LC349894.1) and *Bacillus mycoides* (LC349897.1), ABC- *C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *P. azotoformans* (LC349894.1) and *Bacillus mycoides* (LC349897.1), ABC- *C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *P. azotoformans* (LC349894.1) and *Bacillus mycoides* (LC349897.1), ABC- *C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *P. azotoformans* (LC349894.1) and

consortia of Alcaligenes faecalis (LC349889.1), Pseudomonas azotoformans (LC349894.1) and Bacillus mycoides (LC349897.1), Control- C. olitorius harvested from sterilized soil alone. There was significant difference at p $\leq$ 0.05 in the plant stem diameter of harvested C. Olitorius starting from 2WAP till 7WAP except at 3WAP as shown in Figure 6. By 7WAP, the highest stem diameter was observed in C. Olitorius harvested from soil bioremediated with a consortia of Alcaligenes faecalis (LC349889.1), Pseudomonas azotoformans (LC349894.1) and Bacillus mycoides (LC349897.1) which had an average of  $1.35\pm0.00$  mm while the smallest stem diameter was observed in C. Olitorius harvested from soil bioremediated with Pseudomonas azotoformans (LC349894.1), mixed culture of Alcaligenes faecalis (LC349889.1) and Pseudomonas azotoformans (LC349894.1), mixed culture of Alcaligenes faecalis (LC349889.1) and Bacillus mycoides (LC349897.1) and mixed culture of Pseudomonas azotoformans (LC349889.1) and Bacillus mycoides (LC349897.1) with each having an average of  $1.20\pm0.00$  mm each.



**Keys:** A- *C. olitorius* harvested from sterilized soil bioremediated with *Alcaligenes faecalis* (LC349889.1), B- *C. olitorius* harvested from sterilized soil bioremediated with *Pseudomonas azotoformans* (LC349894.1), C- *C. olitorius* harvested from sterilized soil bioremediated with *Bacillus mycoides* (LC349897.1), AB- *C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *Alcaligenes faecalis* (LC349889.1) and *P. azotoformans* (LC349894.1), AC- *C. olitorius* harvested from sterilized soil bioremediated from sterilized soil bioremediated with mixed culture of *Alcaligenes faecalis* (LC349889.1) and *P. azotoformans* (LC349894.1), AC- *C. olitorius* harvested from sterilized soil bioremediated from sterilized soil bioremediated with mixed culture of *Alcaligenes faecalis* (LC349897.1), BC- *C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *Alcaligenes faecalis* (LC349897.1), BC- *C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *P. azotoformans* (LC349894.1) and *Bacillus mycoides* (LC349897.1), ABC- *C. olitorius* harvested from sterilized soil bioremediated with consortia of *Alcaligenes faecalis* (LC349889.1), *Pseudomonas azotoformans* (LC349894.1) and *Bacillus mycoides* (LC349897.1), Control- *C. olitorius* harvested from sterilized soil bioremediated with consortia of *Alcaligenes faecalis* (LC349897.1), Control- *C. olitorius* harvested from sterilized soil alone.

Tables 2 and 3 shows the result obtained for the post-bioremediation and harvest analysis harvested of proximate matter and heavy metals content of *C. olitorius* respectively. It was observed that there were significant differences in the responses of the various treatment groups.

	Proximate	matter						
Treatment	Fresh weight (g)	Dry weight (g)	% Moisture	Dry matter	% Crude protein	% Ash	% Ether extract (fat)	% Crude fibre
А	0.22±0.05°	$0.10{\pm}0.01^{\rm f}$	55.00±7.07 <sup>g</sup>	$45.00{\pm}7.07^{h}$	$12.24{\pm}1.90^{i,j}$	19.06±1.47°	1.40±0.15 <sup>s</sup>	22.62±0.55t
В	0.26±0.03°	$0.10 \pm 0.03^{f}$	61.90±6.79 <sup>g</sup>	38.10±6.79 <sup>h</sup>	$7.92 \pm 1.17^{jkl}$	20.08±0.11 <sup>no</sup>	1.66±00.22 <sup>rs</sup>	19.78±0.53 <sup>vw</sup>
С	0.36±0.09°	$0.14{\pm}0.04^{\rm f}$	$62.00\pm0.14^{g}$	$38.00{\pm}0.14^{h}$	$15.82{\pm}1.90^{i}$	14.97±0.69 <sup>p</sup>	$1.93 \pm 0.25^{qr}$	18.72±0.25 <sup>w</sup>
AB	0.33±0.13°	$0.15{\pm}0.08^{f}$	$56.25{\pm}8.84^{g}$	$43.75{\pm}8.84^h$	$10.39 \pm 3.18^{j,k}$	17.88±1.62°	$1.91{\pm}0.15^{\mathrm{qr}}$	19.62±0.54 <sup>vw</sup>
AC	0.38±0.05°	$0.14{\pm}0.01^{\rm f}$	61.10±3.25 <sup>g</sup>	$38.90{\pm}3.25^{\rm h}$	$11.03 \pm 2.08^{j}$	18.16±1.48°	1.35±0.22 <sup>s</sup>	18.95±0.96 <sup>vw</sup>
BC	0.29±0.07°	$0.13{\pm}0.04^{\rm f}$	51.95±26.36 <sup>g</sup>	$48.05{\pm}26.38^{h}$	$6.58{\pm}0.03^{k,l}$	$21.65{\pm}0.78^n$	$2.00{\pm}0.05^{qr}$	19.88±0.47 <sup>vw</sup>
ABC	2.54±0.57ª	$0.99 \pm 0.17^{d}$	$60.75{\pm}2.05^{g}$	$39.25{\pm}2.05^{h}$	$4.44 \pm 1.33^{1}$	$24.57{\pm}0.62^m$	$2.00{\pm}0.20^{q,r}$	$21.72{\pm}0.99^{tu}$
Control	1.10±0.13 <sup>b</sup>	0.63±0.24 <sup>e</sup>	43.65±15.34 <sup>g</sup>	$56.35{\pm}15.34^{h}$	$5.57{\pm}0.80^l$	$22.56\pm0.63^{m}$	$2.11 \pm 0.06^{q}$	$20.61 \pm 0.78^{uv}$

 Table 2: Proximate matter of harvested C. olitorius

\*\*\*values with the same letters on each column are not significantly different from each other at  $p \le 0.05$ 

Keys: A- *C. olitorius* harvested from sterilized soil bioremediated with *Alcaligenes faecalis* (LC349889.1), B-*C. olitorius* harvested from sterilized soil bioremediated with *Pseudomonas azotoformans* (LC349894.1), C- *C. olitorius* harvested from sterilized soil bioremediated with *Bacillus mycoides* (LC349897.1), AB- *C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *Alcaligenes faecalis* (LC349889.1) and *P. azotoformans* (LC349894.1), AC- *C. olitorius* harvested from sterilized soil bioremediated from sterilized soil bioremediated with mixed culture of *Alcaligenes faecalis* (LC349889.1) and *P. azotoformans* (LC349894.1), AC- *C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *Alcaligenes faecalis* (LC349897.1), BC- *C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *Alcaligenes faecalis* (LC349897.1), BC- *C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *Alcaligenes faecalis* (LC349897.1), BC- *C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *Alcaligenes faecalis* (LC349897.1), BC- *C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *Alcaligenes faecalis* (LC349897.1), BC- *C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *Alcaligenes faecalis* (LC349897.1), ABC- *C. olitorius* harvested from sterilized soil bioremediated with consortia of *Alcaligenes faecalis* (LC349889.1), *Pseudomonas azotoformans* (LC349894.1) and *Bacillus mycoides* (LC349897.1), Control- *C. olitorius* harvested from sterilized soil alone.

	Heavy n	netals					_	
Treatment	Iron	Copper	Zinc	Cadmium	Lead	Cobalt	Chromium	Nickel
А	25.35±0.49 °	36.05±2.05 <sup>d</sup>	17.45±1.34 <sup>g</sup>	$35.55\pm8.56^{i}$	37.70±0.851	30.50±1.56 <sup>n</sup>	27.00±0.71 <sup>p</sup>	29.00±13.29s
В	26.95±0.64 <sup>bc</sup>	31.70±2.26 <sup>de</sup>	16.50±0.85 <sup>g</sup>	$36.45 \pm 4.60^{i}$	37.65±5.021	27.70±1.41 <sup>n</sup>	$20.75 \pm 1.48^{q}$	29.00±14.57s
С	28.40±1.70 <sup>b</sup>	29.15±1.34 <sup>de</sup>	17.00±0.28 <sup>g</sup>	$32.95 \pm 6.43^{i}$	$34.55 \pm 2.47^{1}$	2835±6.86 <sup>n</sup>	24.80±3.82 <sup>pq</sup>	26.70±8.63 <sup>s,t</sup>
AB	24.95±1.77°	30.05±3.60 <sup>de</sup>	16.10±0.85 <sup>g</sup>	$34.85 \pm 8.13^{i}$	36.30±4.81 <sup>1</sup>	29.35±1.06 <sup>n</sup>	24.40±2.97 <sup>pq</sup>	29.35±11.52 <sup>s</sup>
AC	28.35±1.77 <sup>b</sup>	30.55±6.01 <sup>de</sup>	15.90±0.71 <sup>g</sup>	27.85±11.67 <sup>i,j</sup>	$32.50 \pm 2.55^{1}$	$28.55 \pm 1.48^{n}$	26.40±2.40 <sup>p</sup>	27.70±9.76 <sup>s,t</sup>
BC	28.45±049 <sup>b</sup>	31.00±3.68 <sup>de</sup>	16.10±0.85 <sup>g</sup>	12.00±12.02 <sup>jk</sup>	$31.85 \pm 1.77^{1}$	26.80±0.99 <sup>n</sup>	26.50±2.55 <sup>p</sup>	31.25±11.10 <sup>s</sup>
ABC	37.95±1.20 <sup>a</sup>	$20.20 \pm 4.95^{f}$	4.25±0.21 <sup>h</sup>	$4.40 \pm 0.28^{k}$	10.90±0.85 <sup>m</sup>	6.30±0.71°	4.15±0.64 <sup>r</sup>	$3.20{\pm}0.85^{t}$
Control	38.10±0.85ª	25.80±0.85 <sup>ef</sup>	$5.95{\pm}0.78^{h}$	$4.80{\pm}0.71^{k}$	9.15±0.64 <sup>m</sup>	5.90±0.71°	4.00±0.14 <sup>r</sup>	3.25±0.63 <sup>t</sup>

Table 3: Heavy metal content of harvested C. Olitorius

Keys: A- *C. olitorius* harvested from sterilized soil bioremediated with *Alcaligenes faecalis* (LC349889.1), B- *C. olitorius* harvested from sterilized soil bioremediated with *Pseudomonas azotoformans* (LC349894.1), C- *C. olitorius* harvested from sterilized soil bioremediated with *Bacillus mycoides* (LC349897.1), AB- *C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *Alcaligenes faecalis* (LC349889.1) and *P. azotoformans* (LC349894.1), AC- *C. olitorius* harvested from sterilized soil bioremediated from sterilized soil bioremediated with mixed culture of *Alcaligenes faecalis* (LC349897.1), BC- *C. olitorius* harvested from sterilized from sterilized soil bioremediated with mixed culture of *Alcaligenes faecalis* (LC349897.1), BC- *C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *Alcaligenes faecalis* (LC349897.1), BC- *C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *Alcaligenes faecalis* (LC349897.1), BC- *C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *Alcaligenes faecalis* (LC349897.1), BC- *C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *Alcaligenes faecalis* (LC349897.1), BC- *C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *Alcaligenes faecalis* (LC349897.1), ABC- *C. olitorius* harvested from sterilized soil bioremediated with consortia of *Alcaligenes faecalis* (LC349889.1), *Pseudomonas azotoformans* (LC349894.1) and *Bacillus mycoides* (LC349897.1), Control- *C. olitorius* harvested from sterilized soil bioremediated with consortia of alcaligenes faecalis (LC349889.1), *Pseudomonas azotoformans* (LC349894.1) and *Bacillus mycoides* (LC349897.1), Control- *C. olitorius* harvested from sterilized soil alone.

Table 4 shows the result obtained for heavy metal content of the soil after bioremediation and harvesting of *C. Olitorius* have been conducted. It was observed that the concentration of heavy metals present in the soil was significantly reduced compared to the concentrations observed in the contaminated soil during rainy and dry seasons.

<sup>\*\*\*</sup>values with the same letters on each column are not significantly different from each other at p≤0.05

Parameters	Α	В	С	AB	AC	BC	ABC	CONTROL
Ph	8.50	8.41	8.52	8.49	8.50	8.41	8.51	8.44
	0.40		0 = 0	0.40	0 70		0.40	0.40
Exchangeable	0.40	0.30	0.50	0.40	0.50	0.30	0.40	0.40
Acidity								
(meq/100g)								
Mineral content								
T.O.C (g/kg)	63.52	59.98	65.11	60.74	61.93	67.89	67.09	63.92
T/N (g/kg)	6.57	6.20	6.74	6.28	6.41	7.02	6.94	6.61
Available								
Phosphorus	42.87	33.20	66.48	39.30	39.78	31.46	25.70	49.67
(mg/kg)								
Ca (Cmol/kg)	86.20	105.66	128.11	107.41	99.80	125.62	94.84	101.42
K (Cmol/kg)	2.75	2.78	14.32	2.06	2.42	1.99	0.97	6.04
Na (Cmol/kg)	8.04	8.70	12.39	10.22	9.13	8.91	9.35	8.91
Mg (Cmol/kg)	0.53	1.17	1.08	1.17	0.86	1.23	0.90	1.19
Heavy metals								
(mg/kg)								
Mn (mg/kg)	450.0	515.0	738.0	867.0	581.0	756.0	558.0	910.0
Fe (mg/kg)	17.4	13.5	17.2	15.2	10.2	12.1	14.1	23.5
Cu (mg/kg)	4.12	1.61	2.04	0.97	1.07	2.01	1.51	5.84
Zn (mg/kg)	3.84	1.04	1.68	1.12	0.93	1.68	0.99	4.92
Cd (mg/kg)	2.2	2.6	2.6	2.2	1.8	2.1	2.0	4.8
Pb (mg/kg)	21.3	20.8	40.5	24.3	21.1	20.8	22.6	1360.56
Co (mg/kg)	1.04	1.51	1.68	1.86	1.04	1.33	1.75	8.72
Ni (mg/kg)	2.23	1.06	0.93	1.22	0.86	1.04	1.43	1.21
Cr (mg/kg)	1.96	1.81	3.36	3.04	2.17	2.23	3.06	1240.87

Table 4: Physical and chemical properties of bioremediated soil after harvesting C. olitorius

Keys: A- *C. olitorius* harvested from sterilized soil bioremediated with *Alcaligenes faecalis* (LC349889.1), B- *C. olitorius* harvested from sterilized soil bioremediated with *Pseudomonas azotoformans* (LC349894.1), C- *C. olitorius* harvested from sterilized soil bioremediated with *Bacillus mycoides* (LC349897.1), AB- *C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *Alcaligenes faecalis* (LC349889.1) and *P. azotoformans* (LC349894.1), AC-*C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *Alcaligenes faecalis* (LC349889.1) and *P. azotoformans* (LC349894.1), AC-*C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *Alcaligenes faecalis* (LC349897.1), BC- *C. olitorius* harvested from sterilized soil bioremediated with mixed culture of *P. azotoformans* (LC349894.1) and *Bacillus mycoides* (LC349897.1), ABC- *C. olitorius* harvested from sterilized soil bioremediated with consortia of *Alcaligenes faecalis* (LC349889.1), *Pseudomonas azotoformans* (LC349894.1) and *Bacillus mycoides* (LC349897.1), Control- *C. olitorius* harvested from sterilized soil bioremediated with consortia of *Alcaligenes faecalis* (LC349897.1), Control- *C. olitorius* harvested from sterilized soil alone.

# Discussion

The area surrounding the study site was heavily contaminated with heavy metals as a result of the release of effluent by a steel rolling industry in the area into the surrounding soil. It was observed that the soil could not support plant growth and some plants which could adapt to the heavy metal in the environment were either drying up or had a yellow colour; all these deviations from normal

physical characteristics of a healthy soil and plant indicated a high degree of pollution of the study area by heavy metals. This is in agreement with the report of Revathi et al. (2011), who reported that plant growth is not sustained in soils heavily polluted with heavy metals. The observed deviations in the physical characteristics of the soil samples were in agreement with changes observed in a soil contaminated by acid mine drainage water as reported by Bitala et al. (2009).

Analysis of the contaminated soil sample done before treatment revealed that the concentration of the heavy metals were higher during the dry season than during the wet or rainy season. This can be attributed to the effect of leaching which is more evident during the wet or rainy season compared to the dry season and also the rates of deposition of suspended particles are generally higher during the dry season compared to the wet or rainy season. This is in agreement with the findings of Kilicel (1999).

Concentrations of some heavy metals in the contaminated soil were found to exceed the concentrations recommended by WHO and USEPA. For instance, heavy metals such as cadmium, lead, chromium and cobalt had concentrations of 0.50-2.98 mg/kg, 1505.50-2333.55 mg/kg, 1526.00-1678.67 mg/kg and 10.50-13.65 mg/kg, respectively which exceeded the recommended concentrations of 0.003, 10, 2, and 8 mg/kg by WHO and USEPA for each of the respective heavy metals (Parizanganeh et al., 2012; Ezejiofor et al., 2013); whereas minerals such as calcium, potassium, magnesium and sodium had a concentration of 2.56-2.68, 0.34-0.36, 0.41-0.50 and 0.30-0.50 cmol/kg respectively which were below the recommended values of 10-20, 0.6-1.2, 3-8 and 0.7-1.2 cmol/kg (Parizanganeh et al., 2012; Ezejiofor et al., 2013). These minerals are essential to plant growth and development, a reduction in their bioavailability often leads to reduced plant growth. A major reason for the reduction in bioavailability of these important minerals could be attributed to the high contamination of the soil with heavy metals. According to Chibuike and Obiora (2014), the presence of heavy metals in a soil may affect the availability of other element especially the minerals in the soil.

The result of the molecular characterization of the isolates used in this study revealed high similarity between the isolated organisms and other related organisms in their genus as revealed by the evolutionary tree. The pH of the treated soil obtained in this study ranged between 8.41 and 8.52, this was suitable for the growth and yield of C. olitorius. According to Facciola (1990), C. Olitorius is able to grow well in acid, neutral and basic (alkaline) soils; it tolerates soil pH of 4.5 to 8.0. However, extreme pH conditions have the tendency to reduce the availability of iron in the soil which can cause yellowing between leaf veins (Palada and Chang, 2003).

A mean plant height of 7.84 cm was observed in the group treated with the consortia of was Alkaligenes faecalis (LC349889.1), Pseudomonas azotoformans (LC349894.1) and Bacillus mycoides (LC349897.1), at 7 WAP, this height low compared to that obtained by Ogunrinde and Fasimirin (2011) in which they obtained a mean height of 105.03 cm at 7WAP for C. Olitorius planted in an uncontaminated soil. The stunted growth of plants observed in this study may be as a result of the effect of heavy metals contamination on the soil as indicated by Kabir et al. (2009). Adenipekun et al. (2013) reported similar observation in a study in which they observed a height range of 4.083-11.183 cm in C. Olitorius grown on a soil contaminated with oil and remediated using *Pleurotus pulmonarius* at 5WAP.

A mean of 5.80 was obtained for the for number of leaves at 7 WAP in the group treated with the consortia of Alkaligenes faecalis (LC349889.1), Pseudomonas azotoformans (LC349894.1) and The Proceedings of the Nigerian Academy of Science Volume 13, No 1, 2020

*Bacillus mycoides* (LC349897.1), this is similar to that obtained by Adenipekun *et al.* (2013), in which they observed a range of 4.500-10.666 number of leaves in *C. Olitorius* grown on a *Pleurotus pulmonarius* remediated soil contaminated with oil at 5WAP.

The biggest stem diameter observed among the different treatment groups in this study was found in the group treated with the consortia of *Alkaligenes faecalis* (LC349889.1), *Pseudomonas azotoformans* (LC349894.1) and *Bacillus mycoides* (LC349897.1) having a mean diameter of 1.35 mm. This is comparable to that obtained by Adenipekun *et al.* (2013), in which they observed a range of 0.128-1.05 mm for stem diameter of *C. Olitorius* grown on a *Pleurotus pulmonarius* remediated soil contaminated with oil at 5WAP.

Analysis of the harvested C. olitorius plants revealed that for proximate matter such has % crude fibre, ether extract (fat) and fresh weight most of the treatment groups performed well having a range of 18.72-22.62; 1.35-2.00 and 0.22-2.54 respectively compared to values of 20.30; 0.12 and 1.11 respectively obtained in some previous studies (Adenipekun et al., 2013, Ndlovu and Afolayan, 2008; Yekeen et al., 2013). Other proximate matters such as % ash, % crude protein and % moisture had a range of 14.97-24.57; 4.44-15.82 and 43.65-62.00 respectively and these were low compared to values of 21-21.40; 21.12 and 84.28 respectively obtained by Onwordi et al. (2009) and Acho et al. (2014). High ash content in food is a measure of high deposit of mineral contents (Akpabio et al., 2012). The value obtained in this study suggests that the C. Olitorius harvested was moderately rich in mineral elements. One major reason why the harvested C. Olitorius is not high in fibre content is because the harvested plants were observed to absorb heavy metals and this could have affected its mineral content. Dietary proteins are important for natural synthesis and maintenance of body tissues, enzymes and hormones as well as other substances required for healthy functioning of the body system (Hayat et al., 2014). The protein value obtained from this study 4.44±1.33 to 15.82±1.90 suggests that C. olitorius can effectively contribute to the daily protein needed. Even though the value is low compared to protein values from other reports, Gqaza et al. (2013) however said that any plant food that provides more than 12 % of its caloric value from protein is considered a good source of protein. The moisture content obtained for C. Olitorius in this study though low when compared to that obtained by some previous authors, shows that the harvested C. olitorius can be easily susceptible to spoilage by micro-organisms during storage (George, 2003).

The analysis of the harvested *C. Olitorius* plant revealed the presence of heavy metals in the leaves; this is an indication that there was an uptake of these heavy metals by the roots of *C. olitorius* and this was followed by its translocation which is in agreement with Peralta-Videa *et al.* (2002). According to earlier studies by Kashem and Singh (2004) and Rieuwerts *et al.* (2006), it was reported that at pH ranges between 4.0-8.5, metal cations are mobile while anions tend to transform to oxide minerals, thus increasing their concentration in the environment, the pH of the treated soil samples were found to be between 8.41 and 8.52, this could be one of the reasons the heavy metals were easily absorbed and translocated in the plant. According to Muhammad *et al.* (2008), leafy vegetables grown in heavy metal contaminated soils, accumulate higher amounts of metals than those grown in uncontaminated soils. This is because they are capable of absorbing these metals in their edible and non-edible parts as can be seen in the accumulation of heavy metals in the leaves and stems of *C. Olitorius* in this study. The analysis of the harvested *C. Olitorius* plant revealed that the content of iron (Fe) and zinc (Zn) in the harvested plants was low having a range of 24.95-

38.10 and 4.25-17.45 mg/kg respectively compared to the recommended standard of 60 mg/kg (Sanyaolu *et al.*, 2011; Ayejuyo *et al.*, 2014).

The post bioremediation and post-harvest analysis of the treated soils revealed an increase in the pH, total organic carbon, total nitrogen and available phosphorus of the treated soils. The pH and available phosphorus observed in the treated soil in this study were higher than the pH of 5.1-6.5 and available phosphorus of 20 mg/kg observed in studies by Brady and Weil (2008) and Holland et al. (1981) respectively. An increase was observed in the concentration of minerals such as calcium, potassium and sodium of the treated soils. The isolates used in this study were able to achieve high reduction in heavy metal concentration in the soil. For instance, there was 57.99% reduction in the concentration of iron in the group treated with a mixed culture of Alcaligenes faecalis (LC349889.1) and Bacillus mycoides (LC349897.1), groups treated with Pseudomonas azotoformans (LC349894.1) and Bacillus mycoides (LC349897.1) and a mixed culture of Pseudomonas azotoformans (LC349894.1) and Bacillus mycoides (LC349897.1) had a 98.92% reduction on the concentration of lead, groups treated with Alcaligenes faecalis (LC349889.1) and a mixed culture of Alcaligenes faecalis (LC349889.1) and Bacillus mycoides had a 91.39% reduction on the concentration of cobalt, group treated with Bacillus mycoides (LC349897.1) had a 97.42% reduction on the concentration of nickel while the group treated with *Pseudomonas* azotoformans (LC349894.1) had a reduction of 99.89% on the concentration of chromium. This is comparable to results from earlier studies. For instance, Chang et al. (1997) reported a Pseudomonas aeruginosa PU21 (Rip64) strain with a metal uptake efficiency of 80% within 2 days while Roane et al. (2001) reported a Bacillus strain H9 with a metal uptake efficiency of 36% within 48 hours. Magyarosy et al. (2002) also reported a Pseudomonas spp with a metal uptake efficiency of 98% within 4 days. Though this isolates did not work as rapid as those earlier cited, this can be attributed to the fact that soils contaminated with heavy metals are poor in nutrients and bacterial diversity which results in impeded rates of remediation (White et al., 2006).

# Conclusion

This study has shown that stimulation of bacterial isolates with organic amendments such as cow manure slurry/gomeya has great potentials in bioremediation. It has also been observed that the use of bacterial consortia rather than individual microorganism works more effectively in bioremediation of heavy metals.

# **Conflict of Interest**

The authors declare that there is no conflict of interest.

# References

Acho C F, Zoue L T, Akpa E E, Yapo V G, & Niamke S L (2014). Leafy vegetables consumed in Southern Côte d'Ivoire: a source of high value nutrients. *Journal of Animal andPlant Sciences*, 20(3): 3159-3170.

Adedokun O M & Ataga A E (2007). Effects of amendments and bioaugumentation of soil polluted with crude oil, automotive gasoline oil and spent engine oil on the growth of cowpea (*Vigna unguiculata* L. Walp). *Scientific Research Essay*, 2(5): 147-149.

Adenipekun C O, Ayanleye O O, & Oyetunji O J (2013). Bioremediation of soil contaminated by spent diesel oil using *Pleurotus pulmonarius* Fries (Quelet) and its effects on the growth of *Corchorus olitorius* (L). *Journal of Applied Biosciences*,68: 5366-5373.

Ahluwalia S S, & Goyal D (2007). Microbial and plant derived biomass for removal of heavy metals from wastewater. Bioresource Technology, 98 (12), 2243-2257.

Akan J C, Abdulrahman F I, Ogugbuaja V O, & Ayodele J T (2009). Heavy metals and anion levels in some samples of vegetables grown within the vicinity of Challawa industrial Area, Kano State, Nigeria. American Journal of Applied Sciences, 6(3): 534-542.

Akpabio U D, Udo U E, & Akpakpan A E (2012). Proximate composition and nutrient analysis of Aneilema aequinoctiale leaves. Asian Journal of Plant Science and Research2(5): 607-612.

Aluko O A, Olanipekun T O, Olasoji J O, Abiola I O, Adeniyan O N, Olanipekun S O, Omenna E C, Kareem K O, & Douglas A I (2014). Effect of organic and inorganic fertilizer on the yield and nutrient composition of jute mallow. Global Journal of Agriculture Research, 2(3): 1-9.

Association of Analytical Chemists (AOAC). (2012). Official Methods of Analysis, 19th Ed. Association of Analytical Chemists, Washington D.C.

Ayejuyo O O, Osundiya M O, Olowu R A, Bamgboye O A, & Ogunlola A O (2014). Bioaccumulation of heavy metals in frequently consumed leafy vegetable grown along Nigeria-Benin Seme Border, West Africa. Advances in Applied Science Research, 5(1):1-7.

Ayotamuno J M, Okparanma R N, & Araka P P (2009). Bioaugmentation and composting of oilfield drill-cuttings containing polycyclic aromatic hydrocarbons (PAHs). Journal Food, Agriculture Environment, 7(2): 658-664.

Bates R G (1954). *Electrometric pH determinations*. John Wiley and Sons, Inc. New York. Bitala M F, Kweyunga C, & Manoko M L K (2009). Levels of heavy metals and cyanide in soil, sediment and water from the vicinity of North Mara Gold Mine in Tarime District, Tanzania. A report presented to CCT.

Brady N C, & Weil R R (2008). The Nature and Properties of Soils. 14 ed. Pearson-Prentice Hall, Upper Saddle River, NJ. 990 pp. ISBN: 13-978-0-13-227938-3.

Camargo F A, Okeke B C, Bento F M, & Frankenberger W T (2003). In vitro reduction of hexavalent chromium by a cell-free extract of *Bacillus* sp. ES 29 stimulated by Cu<sup>2+</sup>. Applied Microbiology and Biotechnology, 62(5-6):569-573.

Chang J O, Law R, & Chang C C (1997). Biosorption of lead, copper and cadmium by biomass of Pseudomonas aeruginosa PU21. Water Research. 31:1651–1658

Chibuike G U & Obiora S C (2014). Heavy Metal Polluted Soils: Effect on Plants and Bioremediation Methods. Applied and Environmental Soil Science. http://dx.doi.org/10.1155/2014/752708.

Congeevaram S, Dhanarani S, Park J, Dexilin M, & Thamaraiselvi K (2007). Biosorption of chromium and nickel by heavy metal resistant fungal and bacterial isolates. Journal of Hazardous Materials, 146: 270–277. The Proceedings of the Nigerian Academy of Science Volume 13, No 1, 2020

Dixit R, Malaviya W D, Pandiyan K, Singh U B, Sahu A, Shukla R, Singh B P, Rai J P, Sharma P K, Lade H, & Paul D (2015). Bioremediation of Heavy Metals from Soil and Aquatic Environment: An Overview of Principles and Criteria of Fundamental Processes. Sustainability, 7:2189-2212.

Duruibe J O, Ogwuegbu M O C, & Egwurugwu J N (2007). Heavy metal pollution and human biotoxic effects. International Journal of Physical Sciences, 2(5):112-118.

Eddy N O (2004a). Physicochemical parameter of water and heavy metal content of water, sediment and fishes from Qua Iboe River Estuary. M.Sc Thesis. Michael. Okpara University of Agriculture, Umudike. Nigeria.

Elings A (2000). Estimation of leaf area in tropical maize. Agronomy Journal, 92:436 – 444. Ernst W H O (1998). The origin and ecology of contaminated, stabilized and non-pristine soils. In: Metal-contaminated soil. Vangronsveld, J. and Cunningham, S.D. (Eds). (Chapter 2). Springer, New York, pp. 17-29.

Ezejiofor T I N, Ezejiofor A N, Udebuani A C, Ezeji E U, Ayalogbu E A, Azuwuike C O, Adjero I A, Chinedu I, Cosmas U O, Nwaogu I A, &Ngwogu K (2013). Environmental metals contaminants load of a densely populated and heavily industrialized commercial city of Aba, Nigeria. Journal of Toxicology and Environmental Health Sciences, 5(1): 1-11.

Facciola S C (1990). A source book of edible plants. Kampony Publications. pp. 10-15. Federal Remediation Technologies Roundtable (FRTR) (2000). In-Situ Biological Treatment. technologies Remediation screening matrix and reference guide, Version 4.0.www.frtr.gov/matrix2/section4/4\_1.html.2004/04/07.

George P M (2003). Encyclopedia of foods. Humane Press, Washington D.C, 1: 526. Gomes H I, Dias-Ferreira C, & Ribeiro A B (2012). Electrokinetic remediation of organochlorines in soil: Enhancement techniques and integration with other remediation technologies, Chemosphere, 87:1077-1090.

Green-Ruiz C, Rodriguez-Tirado V, & Gomez-Gil B (2008). Cadmium and zinc removal from aqueous solutions by Bacillus jeotgali: pH, salinity and temperature effects. Bioresource Technology, 99: 3864-3870.

Gqaza M B, Njume C, Goduka I N, & Grace G (2013). The proximate composition of S. nigrum plant-leaves consumed in the Eastern Cape Province of South Africa. International Conference on Nutrition and Food Sciences. DOI:10.7763/IPCBEE.

Guha H, Jayachandran K, & Maurrasse F (2001). Kinetics of chromium (VI) reduction by a type strain Shewanella alga under different growth conditions. Environmental Pollution, 115:209–218. Hall B G (2013). Building Phylogenetic Trees from Molecular Data with MEGA. Molecular Biology and Evolution (mbe.oxfordjournals.org). 30 (5): 1229-1235.

Hayat I, Ahmad A, Ahmed A, Khalil S, & Gulfraz M (2014). Exploring the potential of red kidney beans (Phaseolus vulgaris L.) to develop protein based product for food applications. Journal of Animal and Plant Sciences, 24(3): 860-868. The Proceedings of the Nigerian Academy of Science Volume 13, No 1, 2020 48

Holland M D, Allen R K G, Barten D, & Murphy S T (1989). Land Evaluation and Agricultural for Cross River National Park, Oban Division. Prepared by the Overseas Development National Resources Institute in Collaboration with WWF for the Federal Republic of Nigeria and the Cross River State government, 1189.

Hussein H, Krull R, Abou El-Ela S I, & Hempel D C (2001). *Interaction of the different heavy metal ions with immobilized bacterial culture degrading xenobiotic wastewater compounds*. In Proceedings of the Second International Water Association World Water Conference, Berlin, Germany, 15–19 October; pp. 15–19.

Iqbal M A, Chaudhary M N, Zaib Z, Imran M, Ali K, & Iqbal A (2011). Accumulation of heavy metals (Ni, Cu, Cd, Cr, Pb) in agricultural soils and spring seasonal plants, irrigated by industrial waste water. *Journal of Environmental Technology and Management*, 2(1): 1554-2010.

Kabir M, Iqbal M Z, & Shafiq M (2009). Effects of Lead on Seedling Growth of *Thespesia* populnea. Advances in Environmental Biology, 3(2): 184-190.

Kashem M A, & Singh B R (2004). Transformation in solid phase species of metals as affected by flooding and organic matter additions in contaminated soils. *Communications in Soil Science and Plant Analysis*, 35: 1435-1456.

Kilicel F (1999). Investigation of toxic heavy metals pollution in road dust at the center of Van Turkey. *Bulletin of Pure and Applied Science*, 18: 1-4.

Magyarosy A, Laidlaw R D, Kilaas R, Echer C, Clark D, & Keasling J D (2002). Nickel accumulation and nickel oxalate precipitation by *Aspergillus niger*. *Applied Microbiology and Biotechnology*, 59: 382-388.

Mclean E O (1965). *Aluminium*. In C.A. Black (ed), Methods of Soil Analysis. Part 2, pp 986-994.

Mehlich A (1953). Determination of P, Ca, Mg, K, Na, and NH4. North Carolina Soil Test Division.

Muhammad F, Farooq A, & Umar R (2008). Appraisal of heavy metal contents in different vegetables grown in the vicinity of an industrial area. *Pakistan Journal of Botany*,40(5): 2099-2106.

Narasimhulu K, Rao P S, & Vinod A V (2010). Isolation and identification of bacterials strains and study of their resistance to heavy metals and antibiotics. *Journal of Bacterial and Biochemical Technology*, 2: 074-076.

Navarro M C, Pérez-Sirvent C, Martínez-Sánchez M J, Vidal J, Tovar P J, & Bech J (2008). Abandoned mine sites as a source of contamination by heavy metals: A case study in a semi-arid zone. *Journal of Geochemical Exploration*, 96(2–3):183–193.

Ndlovu J, & Afolayan A J (2008). Nutritional analysis of South African wild vegetable *Corchorus olitorius* L. *Asian Journal of Plant Sciences*, 7(6): 615-618.

Odoh R & Kolawole S A (2011). Assessment of trace heavy metal contaminations of some selected vegetables irrigated with water from River Benue within Makurdi Metropolis, Benue State Nigeria. *Advances in Applied Science Research*, 2(5):590-601.

Ogunrinde A T & Fasinmirin J T (2011). Soil moisture distribution pattern and yield of jute mallow (*Corchorus olitorius*) under three different soil fertility management. *Proceedings of the Environmental Management Conference*, Federal University of Agriculture, Abeokuta, Nigeria. pp 373-381.

Okparanma R N, Ayotamuno J M, & Araka P P (2009). Bioremediation of hydrocarbon contaminated-oil field drill-cuttings with bacterial isolates. *African Journal of Environmental Science and Technology*, 3(5):131-140.

Olutiola P O, Famurewa O, & Sonntag H G (2000).*An introduction to general microbiology*: A practical approach, Hygiene Institute DerUniversitatHeldelberg.pp 157-180.

Onwordi C T, Ogungbade A M, & Wusu A D (2009). The proximate and mineral composition of three leafy vegetables commonly consumed in Lagos, Nigeria. *African Journal of Pure and Applied Chemistry*, 3(6): 102-107

Özdemir S & Kılınc E (2012). *Geobacillusthermoleovoran s*immobilized on Amberlite XAD-4 resin as a biosorbent for solid phase extraction of uranium (VI) prior to its spectrophotometric determination. *Microchimica Acta*, 178:389–397.

Özdemir S, Kılınc E, Poli A, & Nicolaus B (2013). Biosorption of Heavy Metals (Cd<sup>2+</sup>, Cu<sup>2+</sup>, Co<sup>2+</sup>, and  $Mn^{2+}$ ) by Thermophilic Bacteria, *Geobacillus thermantarcticus* and *Anoxybacillus amylolyticus*: Equilibrium and Kinetic Studies. *Bioremediation Journal*, 17:2, 86-96.

Page A L (eds) (1982). *Methods of soil Analysis Agronomy*. American Society of Agronomy and Soil Science, Madison, Wisconsin. No. 9. Part 2.

Palada M C & Chang L C (2003). Suggested Cultural Practices for Jute Mallow. *Asian Vegetable Research & Development Council Centre (AVRDC).* 

Parizanganeh A H, Bijnavand V, Zamani A A, & Hajabolfath A. (2012). Concentration, Distribution and Comparison of Total and Bioavailable Heavy Metals in Top Soils of Bonab District in Zanjan Province. *Open Journal of Soil Science*, 2: 123-132.

Perarlta-Videa J R, Gardea-Torresdey J I, Gomez E, Tiemann K J, Parsons J G, & Carrillo G (2002). Effect of mixed cadmium, copper, nickel and zine at different PH upon alfalla growth and heavy metal uptake. *Environmental Pollution*, 119: 291-302.

Randhawa G R & Kullar J S (2011). Bioremediation of pharmaceuticals, pesticides, and petrochemicals with gomeya/Cattle dung. *International Scholarly Research Network. ISRN Pharmacology.* doi:10.5402/2011/362459. The Proceedings of the Nigerian Academy of Science

Revathi K, Haribabu T E, & Sudha P N (2011). Phytoremediation of Chromium contaminated soil using Sorghum plant. *International Journal of Environmental Sciences*, 2(2): 417-428.

Rieuwerts J S, Ashnore M R, Farago M E, & Thornton I (2006). The influence of soil characteristics on the extractability of Cd, Pb and Zn in upland and moorland soils. *Science of the Total Environment*, 366: 864-875.

Roane T M, Josephson K L, & Pepper I L (2001). Dual-bioaugmentation strategy to enhance remediation of cocontaminated soil. *Applied Environment Microbiology*, 67: 3208-3215.

Saeed G, & Rafique M (1980). Government of Pakistan, Ministry of Food and Agriculture, Soil survey of Pakistan, Lahore. *Technical guide for the chemical analysis of soil and water*, Bulletin No. 14.

Saitou N, &Nei M (1987). The neighbor-joining method: A new method for reconstructing phylogenetic trees. *Molecular Biology and Evolution*. 4: 406-425.

Sanyaolu V T, Sanyaolu A A A, & Fadele E (2011). Spatial variation in Heavy Metal residue in *Corchorus olitorious* cultivated along a Major highway in Ikorodu- Lagos, Nigeria. *Journal of Applied Sciences and Environmental Management*, 15(2): 283-287.

Soil Science Society of America (SSSA) 1971. *Instrumental methods for analysis of soils and plant tissue*. Soil Science Society of America Inc. Madison. Wis. USA. pp. 27-32.

Sukreeyapongse O, Panichsakspatana S, & Hansen H (2002). Transfer of heavy metals from sludge amended soil to vegetables and leachates. Paper presented at the 17<sup>th</sup> World Congress of Soil Science (WCSS), 14<sup>th</sup>-21<sup>st</sup>August, 2002. Thailand. Symposium No. 29, Paper No. 1969.

Sugiyama M (1994). Role of cellular antioxidants in metal-induced damage. *Cell Biology and Toxicology*, 10:1-22.

Suruchi I, & Khanna P (2011). Assessment of Heavy Metal Contamination in Different Vegetables Grown in and Around Urban Areas. *Research Journal of Environmental Toxicology*, 5:162-179. Tamura K, Nei M, & Kumar S (2004). Prospects for inferring very large phylogenies by using the neighbor-joining method. *Proceedings of the National Academy of Sciences (USA)*. 101: 11030-11035.

Tamura K, Stecher G, Peterson D, Filipski A, & Kumar S (2013). MEGA6: Molecular Evolutionary Genetics Analysis version 6.0. *Molecular Biology and Evolution*.30: 2725-2729.

United State Environmental Protection Agency (USEPA) (1995). Bioremediation of hazardous wastes: Research, Development, and Field Evaluations EPA/540/R-95/532.

Watanabe F S & Olsen S R (1965). Test of an ascorbic acid method for determining phosphorus in water and NaHCO<sub>3</sub> extracts from soils. *Soil Science Society of America Proceeding*. 29:677-678.

White P M, Wolf D C, Thomas G J, & Reynolds C M (2006). Phytoremediation of alkylated polycyclic aromatic hydrocarbons in an oil-contaminated soil. *Water Air and Soil Pollution*, 169: 207-220.

Yekeen T A, Akintaro O I, Akinboro A, & Azeez M A (2013). Evaluation of cytogenotoxic and nutrient composition of three commonly consumed vegetables in south-western Nigeria. *African Journal of Food, Agriculture, Nutirtion and Development.* 13(2): 7452-7466.

Yusuf A A, Arowolo T A, & Bamgbose O (2003). Cadmium, copper and nickel levels in vegetables from industrial and residential areas in Lagos City, Nigeria. *Food and Chemical Toxicology*,41:375-378.